

Water Science and Technology Library

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Emerging and Innovative Arsenic Removal Technologies for a Sustainable Future

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Chapter 1

Arsenic in Water: Understanding the Silent Threat



Yusra Sharf and Rushda Sharf

Abstract In Minnesota, arsenic is found naturally in soil and rocks. Volcanoes, the weathering of minerals and ores that contain arsenic, and commercial or industrial processes all release arsenic into the atmosphere. Groundwater that could be used for drinking water may absorb trace amounts. Compared to organic and inorganic arsenic, inorganic arsenic is typically too toxic. In certain nations, inorganic arsenic can be found in private well water, copper-chromated arsenate-treated lumber, and industrial settings. For the vast majority, drinking water contaminated with naturally occurring inorganic arsenic is the source of contact. The biggest risk to public health is drinking water polluted with arsenic, which can increase the risk of cancer and other serious health consequences. The most common ways that arsenic exposure results in disease are through ingestion and inhalation. The symptoms of arsine gas poisoning differ significantly from those of other types of arsenic poisoning. Arsine rapidly linked itself to red blood cells after inhalation, causing irreversible destruction of the cell membrane. Neurological, respiratory, hematologic, cardiovascular, gastrointestinal, and other systems can all be adversely affected by arsenic. Arsenic causes cancer in several of organ systems. Skin, lung, prostate, bladder and liver angiosarcoma, are the most prevalent forms of cancer. The central nervous system, cell respiration, inhibition of the pyruvate dehydrogenase enzyme complex, and overt gastrointestinal disturbances may be caused by acute or subacute exposure to As (more than a few mg of inorganic arsenic/day). Chronic arsenic hyperpigmentation can affect mucous membranes in addition to the trunk, where it manifests itself as a thinly dappled “raindrop” manner. Arsenic is a teratogen and a reproductive toxin. The concentrations in the cord blood and maternal blood are comparable, and it is transferred across the placenta. The chapter focuses on causal inference, which is also known as hazard identification and clinical prevention in the terminology of risk assessment.

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1.1 Introduction

The concentration of arsenic (As) in the Earth's crust is 1.8 parts/million (ppm). It can be found as sulfides (As_2S_3) such as arsenopyrites (FeAsS) rather than in its elemental form. As is harmful in each form. Since it is a redox-sensitive element, pH and redox potential have an impact on its mobility in groundwater (Perkins, 2022). Worldwide, freshwater and marine environments contain arsenic, which poses a threat to aquatic life. This material occurs in both inorganic and organic forms. Variations in As toxicity have been present in monomet hylarsonate (MMA), arsenite, arsenate, and dimethyl arsenate (DMA). Deeper groundwater layers naturally occur at high As levels; this is referred to as "contamination." Globally, the Ganga–Brahmaputra region is the most polluted place on Earth and is a common element in groundwater. Arsenate (H_2AsO_4) and arsenite are the two predominant inorganic forms in water (H_2AsO_3). Arsenic is a critical environmental problem due to its bioaccumulation and movement through the aquatic food chain at different trophic levels. Arsenic bioaccumulates in aquatic organisms, such as fish, after prolonged exposure to low concentrations of the metal. When fish tainted with arsenic are consumed, humans may be exposed to the metal. The total amount of As that the gastrointestinal tract can easily absorb makes up 95% of the inorganic arsenic (iAs) in drinking water. However, up to 50–70% of the absorbed arsenic is removed by methylation in the kidneys, and this is subsequently eliminated through urine. When more As is consumed than eliminated, it tends to build up in tissues, including the hair, nails, and other body parts (Moulick et al., 2021). As is mostly present in the shallow groundwater zone, where clay and sulfite minerals combine to form in a reducing environment. During the summer session, groundwater levels rapidly decreased, leading to the oxidation of sulfides and their aeration into solution. The Worldwide problems associated with drinking contaminated water and groundwater contamination by As have been documented. Pigmentation, keratosis, and skin cancer are believed to be typical signs of chronic toxicity caused by long-term ingestion of tainted water. The most common type of As toxicity in humans is caused by interactions with various biological ligands that inactivate the enzyme system, and the symptoms of As exposure are keratosis in the soles and palms and hyperpigmentation, especially on the trunk. Some studies have reported that these skin lesions frequently appear 5–10 years after the first exposure (Adeloju et al., 2021). Arsenic is one of the most hazardous substances on Earth and is well known to cause cancer. Arsenicosis, is As-related health issues such as skin, bladder, kidney and lung cancer, probably diabetes, elevated blood pressure, and reproductive disorders, can be caused by excessive and prolonged (such as five to ten years) human ingestion of toxic food and drinking water (Fig. 1.1) As is currently thought to impact over 150 million people worldwide due to its progressively higher levels in drinking water (Shirin & Yadav, 2014; Stroud et al., 2011). Major river